Care and Feeding of Rechargeable Batteries

Chris Capener March 1, 2012



- * Lead Acid
- * Nickel-Based
 - * NiCd
 - * NiMH
 - * LSD
- * Li-ion

Battery Charging

* Lead Acid

* Nickel-based

* Battery Packs

* Analyzers & Chargers

Before We Begin

Some Definitions

Battery capacity is rated in Amp-hours (Ah)

- It is the number of amperes times the number of hours that the battery can supply
- A 20 Ah battery can supply 1 A for 20 hours, 4 A for 5 hours, etc.

The letter 'C' is used to represent the capacity of a battery
Often used in describing charge rates
To charge a 1 Ah battery at a 0.5C rate means to charge with 500 mA



Different Chemistries

Sealed Lead Acid (SLA)



Sealed Lead Acid (SLA)

* Pros

- Inexpensive and simple to manufacture
- Mature, reliable and wellunderstood technology
- * Self-discharge is among the lowest of rechargeable batteries
- Capable of high discharge rates

* Cons

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- Low energy density poor weight-toenergy ratio limits use to stationary and wheeled applications.
- Cannot be stored in a discharged condition - the cell voltage should never drop below 2.1 OV.
- Allows only a limited number of full discharge cycles - well suited for standby applications that require only occasional deep discharges.

Nicd - Nickel Cadmium







Nicd - Nickel Cadmium

* Pros

- Fast and simple charge
- High number of charge/discharge cycles over 1000 cycles
- Good load performance
- * Good low temperature performance
- One of the most rugged rechargeable batteries.
- * Economically priced

* Cons

- Relatively low energy density
- Memory effect nickel-cadmium must periodically be exercised (discharge/charge) to prevent memory
- Environmentally unfriendly nickelcadmium contains toxic metals
- Relatively high self-discharge needs recharging after storage

NiMH -Nickel Metal Hydride







NiMH -Nickel Metal Hydride

* Pros

- 30-40% higher capacity than standard nickel-cadmium
- Less prone to memory than nickelcadmium
- Environmentally friendly contains only mild toxins

* Cons

- Limited discharge current heavy load reduces the battery's cycle life.
- * More complex charge algorithm needed
- Trickle charge settings are critical
- High self-discharge typically 50%
 higher than nickel-cadmium
- High maintenance nickel-metal hydride requires regular full discharge to prevent crystalline formation

LSD - Low Self Discharge







Also known as 'Hybrid'

LSD - Low Self Discharge

* Pros (vs. NiMH)

- Much longer shelf life
- * Better cycle life
- * Same environmentally friendliness



* Cons (vs. NiMH)

- * Lower initial capacity
- Higher cost



Days in Storage

- * After only three weeks of storage, the Eneloops have more capacity remaining.
- * After about 3.5 months, the Eneloops will have twice the capacity of the traditional cells.

Cycle Life



Li-ion - Lithium Ion



- Highest energy density available
- * Good cycle life: >500 cycles

* Cons

- * Risk of FIRE if abused
- Complex, tightly controlled charger

High cost

 Were only available packaged with safety circuit - Not anymore

Do not try this at home







Laptop Battery Pack

Laptop Explodes in LAX 6/5/07

Battery Charging

Different Procedure for Each Chemistry

Sealed Lead Acid (SLA)



Stage 1: Constant Current between 0.1C and 0.3C

Stage 2: Constant Voltage at 2.4V/cell (14.4V for a 12V battery) for 5 hours

Stage 3: Float Charge at 2.25V/cell (13.5V for a 12V battery)

Nickel-Based



 Avoid high temperature during charging
 A charger for nickel-metal-hydride can also accommodate nickel cadmium, but not the other way around. A charger designed for nickel-cadmium would overcharge the nickel-metal-hydride battery.
 Nickel-based batteries prefer fast-charge. Lingering slow charges cause crystalline formation (memory).

If not used immediately, remove the battery from the charger and apply a topping-charge before use. Do not leave nickel-based battery in the charger for more than a few days, even if on trickle charge.



And why they often die young

Battery Pack



Made up of multiple cells in series



The upper cell has less capacity than the lower one

During discharge it will empty first

The lower cell still has some capacity left

If discharge is continued, the upper cell will be damaged

Two Cell Example





- * Use a battery pack made from individual rechargeable cells
 - * AA battery pack
- * Charge the cells individually outside of the pack
- * Po not run the pack all the way down

One Issue



* Alkaline cells: $*4 \times 1.5V = 6.0V$ * NiMh cells $*4 \times 1.2V = 4.8V$ * Lower output power! * 1.5W

Need more cells



* Look for 6-cell AA battery cases



* 3.5W



Why to spend a little extra

Considerations



- * Capable of fast charge
- * Voltage and temperature monitoring
- * Capacity analysis

LaCrosse Alpha BC-700



Charging Current: Indicates mA charge rate	200 Displays & Modes
Discharging Current:	A. Charge Mode Within 4 seconds charging automatically begins at 200mA, or select 500, 700, 1000, 1500 or 1800 mA settings for faster charging times.
Indicates mA discharge rate	B. Discharge Mode Reduce memory effect in rechargeable batteries by discharging completely then recharging to full capacity in one cycle.
Time Elapsed: Indicates charging time	221 User selectable mA.
(hh:mm)	C. Refresh Mode Batteries will be discharged and recharged 20 cycles or until batteries are refreshed to
Accumulated Capacity: Indicates battery's full capacity	
(mAh or Ah)	CHARGE TEST D. Test Mode Batteries are charged to full capacity, discharged completely to measure capacity for display in mAh or Ah, then recharged to full capacity.
Terminal Voltage: Measures battery voltage	
(can be displayed in CHAI any mode)	Automatically switches to Trickle Charg at the end of last charging cycle.



For the truly committed

West Mountain Radio CBA-III





Cadex Battery Analyzer



Acknowledgements & Links

- * Battery University
 - * http://www.batteryuniversity.com
- http://www.stefanv.com/electronics/sanyo_eneloop.html
- http://www.eneloop.info
- http://www.lacrossetechnology.com/bc700
- http://www.westmountainradio.com
- http://www.cadex.com